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Chapter 7, Section 7.2 1. How many

valence electrons does a neutral

magnesium (Mg) atom have? 2. What is

the charge on a magnesium ion? What

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does magnesium have to do to form such
an ion, and why does it tend to do so? 3.

How many valence electrons does a single
neutral chlorine (Cl atom) have? 4.

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WORKSHEET The Activity Series Use with Chapter 9, Section 9.2 1. For each of the following pairs of elements, underline the one that would replace the other element in a compound.

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answer to question 5, how many electrons

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clockwise in a p sublevel in the 4th ...

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WORKSHEET Atomic Orbitals 1. What is the shape of an s orbital? Class 16 Use with Chapter 5, Section 5.2 2. What is the relationship between the size of an s orbital and the principal energy level in which it is found? 3. What is the shape of a p orbital? How many p orbitals are there in a ublevel? 4.

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Guide and Answers Teaching

Transparency 22 – Formation of Ions 1.

calcium and oxygen 2. Yes; each contains

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equal numbers of protons and electrons. 3.
ionization energy 4. positive; cation 5.
negative; anion 6. No; it is not a stable
octet of electrons. 7.

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