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Chapter 7. The chemical

formula for an ionic

compound represents

one formula unit - the

simplest ratio of the

compound's positive

ions (cations) and its

negative ions (anions).

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Monatomic cations are identified simply by the element's name.

Monatomic anions more complicated. The ending of the element's name is dropped, the "ide" added to the root name.

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formula. subscript of  
each element reveals  
number of atoms  
contained in a single  
molecule of a certain  
compound. monoatomic  
ions. ions formed from a  
single atom; remember  
to change nonmetals to  
-ide ending. oxyanions.  
polyatomic ions that

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Chapter 7. monatomic ions. binary compounds. nomenclature.

oxyanions. ions formed from a single atom.

compounds composed of two elements. naming system;... name of the cation is given first,

followed by polyatomic ions that contain

oxygen;... an anion that has one few monatomic

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ions. ions formed from a  
single atom. binary  
compounds. compounds  
composed of two ...

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Problem 1 Describe the contributions made by the following scientists to the development of

the periodic table: a.

Stanislao Cannizzaro b.

Dmitri Mendeleev c.

Henry Moseley. HC

Hashim C. Numerade

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Problem 2 State the

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ANSWER Answer the  
Answers

following questions in  
the space provided. 1.

Label each of the

following statements as

True or False: True a. If

the formula mass of one

molecule is  $x$  amu, the

molar mass is  $x$  g/mol.

False b.

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