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CHAPTER 17 REVIEW Reaction Kinetics MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. The reaction for the decomposition of hydrogen peroxide is  $2\text{H}_2\text{O}_2(\text{aq}) \rightarrow 2\text{H}_2\text{O}(\text{l}) + \text{O}_2(\text{g})$ . List three ways to speed up the rate of decomposition. For each one, briefly explain why it is effective, based on collision theory.

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Chapter 10 States of Matter. Educators. LO Chapter Questions. 01:16. Problem 1 What idea is the kinetic-molecular theory based on? ... Problem 17 Explain the relationship between atmospheric pressure and the actual boiling point of a liquid. Rabia B.

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States of Matter | Holt: Modern Chemistry | Numer...  
Holt Modern Chemistry Review CHAPTER 12 & 13:  
SOLUTIONS The following pages contain the bulk (but not all) of the information for the chapter 12 & 13 test. Focus on this content, but make sure to review class notes, activities, handouts, questions, etc.

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