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Heat Capacities The generic heat capacity c

105 bT $C_p = a + \text{ÅÅÅÅÅÅÅÅÅÅÅÅ} +$

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heat capacities of various elements and
compounds are $C_p = a + bT + cT^2$.
a = 21.30, b = 8.54, c = 1.51 ...

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[134wwr85xw47] thermodynamics has much wider applicability. In material science, one is normally not that interested in heat and work, but interested more the state of matter and how things might change when mixed, heated, pressurized, etc.

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DH . Work is found the the first law as $w = q - DU$; thus $q = DH$; $w = DHPVL$; 4.

Isothermal Process Because U is a function only of T for an ideal gas, $DU = DH = 0$ for an isothermal process. These results also follow from the general results

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thus $q = \Delta H$; $w = \Delta H - \Delta U$; 4. Isothermal
Process Because U is a function only of T
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