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Solutions to exercises 1B.1(b) The mean speed is  $\sqrt{8RT/v_{\text{mean}}M}$  The mean translational kinetic energy is  $\frac{3}{2}RT$   $\frac{3}{2}kT$   $\frac{1}{2}Mv_{\text{mean}}^2$  The ratios of species 1 to species 2 at the same temperature are  $\frac{E_{k,1}}{E_{k,2}} = \frac{v_{\text{mean},1}}{v_{\text{mean},2}}$  (i)  $\frac{1}{2}mv^2$   $\frac{1}{2}M_1v_{\text{mean},H}^2$   $\frac{1}{2}mv^2$  and  $\frac{1}{2}M_2v_{\text{mean},Hg}^2$   $\frac{200.6}{4.003} = \frac{1}{2} \frac{m_1v_{\text{rms},1}^2}{m_2v_{\text{rms},2}^2} = \frac{E_{k,1}}{E_{k,2}} = 7.079$  (ii) The mean translation kinetic energy is independent of molecular mass and depends upon temperature alone!

Atkins' Physical Chemistry Instructors Solutions Manual ...

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