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Unit 7 Atoms

\u0026 Molecules

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Periodic

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~~Structure and~~

~~Periodicity~~

Valence

Electrons and

the Periodic

Table

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Properties of

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the Elements:

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Table, elements

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~~heterodiatomic~~

~~Molar volume~~

~~10th SCIENCE~~

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~~(Atomic~~

~~Structure and~~

~~Periodicity)~~

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SECTION 1 SHORT

ANSWER Answer

the following

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questions in the space provided.

1. c In a Stock system name such as iron(III) sulfate, the Roman numeral tells us (a) how many atoms of Fe are in one formula unit.
(b) how many sulfate ions can be attached to

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the iron atom.

(c) the charge
on each Fe ion.

Answers

7 Chemical

Formulas and

Chemical

Compounds

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STUDY. PLAY. The

basis of an

ionic bond is

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the__ absorption
of water into
their solid
structures.

Elements tend to
gain or lose
electrons in
order to acquire
the electron
configuration of
a__ noble gas.

Ions that are
formed from more
than one atom

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are called__

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PLAY. Match.

Gravity. Created
by. dolphinking.
study the thing.

Terms in this
set (25)

Chemical
formula.

indicates the
relative number
of atoms of each
kind in a
chemical
compound.

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Reveals the
number of atoms
of each element
contained in a
single molecule

...

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Name _____ Class _____ Date _____

Chapter Test B,
continued Write
the orbital

notation for the
following

elements in the
space provided

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answer key. 37.

lithium, atomic

number 3 38.

carbon, atomic

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number 6 39

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chapter 7 test
answer key.

neon, atomic

number 10 PART

VI Write the

answers to the

questions on the

line to the

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Answers

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Answers such as

iron(III)

sulfate, the

Roman numeral

tells us (a) how

many atoms of Fe

are in one

formula unit.

(b) how many

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sulfate ions can be attached to the iron atom.

(c) the charge on each Fe ion.

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VOCABULARY

Complete each
sentence. 1.

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numbers are
assigned to the
atoms composing
a compound or
ion in order to

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Answer Key

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Section 3.

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Modern Masses.

any molecule,
formula unit, or
ion is the sum
of the average
atomic masses of
all atoms
represented in
its formula. a
compound's molar
mass is
numerically ...

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Review. 4 terms.

Answers

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3 Review Answers

elements in the

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answer key. 37.

lithium, atomic

number 3 38.

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carbon, atomic
number 6, 39

Modern chemistry
chapter 7 test
answer key.

neon, atomic
number 10 PART

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answers to the
questions on the
line to the
left, and ...

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Answer Key

Chemistry

Modern Chemistry

Chapter 7 Test B

Answers

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Answers As

recognized,

adventure as

capably as

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