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~~Oxidation \u0026~~

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for the following
chemical reactions
How to Predict
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The chemical
reactions that occur
inside the human
body are extremely
complex. One of the
main reactions that

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occurs is the metabolism of glucose, and another reaction is the detoxification of the liver. The metabolism of glucose is one of the main factors that affect the levels of insulin that are generated by the liver.

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Assessment.

Describe the steps in writing a balanced

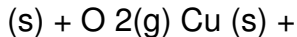
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Chemical equation.

Write the skeleton equation for the following reactions:

Heating solid copper(II)sulfide in the presence of oxygen gas produces pure copper and sulfur dioxide gas. Iron metal and chlorine gas react to form solid iron(III)chloride. CuS



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SO₂(g) D

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chemical equation

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Chemical is a representation of a chemical reaction with reactants on the left, products on the right, and an arrow separating the two.

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Additional Questions

& Answers. Question

1. Give an example of
a complex chemical

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reaction. Answer:

‘Photosynthesis’ is a complex chemical reaction. Question 2.

Give an example of simple chemical reactions. Answer: Rusting of iron and combustion of fuels are simply chemical reactions. Question 3.

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Types of Chemical

Reactions Often

charcoal briquettes

provide the heat for

barbeque grills

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through the burning of carbon. Have you ever felt the heat and smelled the smoke coming from a burning charcoal grill? The heat and smoke are the products of a combustion reaction. Combustion is one of the 4 general types of chemical

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~~oxidation–reduction~~

~~reactions A chemical~~

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a change in the oxidation states of one or more elements in the reactants that has the general form oxidant + reductant ? reduced oxidant + oxidized reductant.

The general forms of these five kinds of reactions are summarized in Table 11.6.1, along with examples of each.

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~~Chapter 11.6: Types
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...

As shown in Figure 11.3.1, applying a small amount of heat to a pile of orange ammonium dichromate powder results in a vigorous reaction known as the ammonium

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dichromate

volcano. Heat, light, and gas are produced as a large pile of fluffy green chromium(III) oxide forms. We can describe this reaction with a chemical equation An expression that gives the identities and quantities of the ...

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30 A halogen can also

replace another

halogen from a

compound. •The

activity of halogens

decreases as you go

down Group 7A of the

periodic

table—fluorine,

chlorine, bromine, and

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iodine. •Bromine is more active than iodine, so this reaction occurs: $\text{Br}_2(\text{aq}) + 2\text{NaI}(\text{aq}) \rightarrow 2\text{NaBr}(\text{aq}) + \text{I}_2(\text{aq})$

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