

Chapter 10 Review Physical Characteristics Of Gases Answer Key

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~~SECTION 10-1. SHORT ANSWERAnswer the following questions in the space provided. 1. Identify whether the descriptions below describe an ideal gas or a real gas. a. Gas particles move in straight lines until they collide with other particles or the walls of their container.~~

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~~Chapter 10 - Physical Characteristics of Gases. 10-1 The Kinetic-Molecular Theory of Matter. I. The Kinetic-Molecular Theory of Gases A. Ideal Gas 1. An imaginary gas that perfectly fits all the assumptions of the kinetic- molecular theory B. Five Assumptions of the Kinetic-Molecular Theory 1. Gases consist of large numbers of tiny particles that are far apart relative to their size 2.~~

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~~Chapter 10 - Physical Characteristics of Gases. 1. ChemistryS. Martinez - Spring 2009 1. 2. KMT - based on idea that particles of matter are always in motion. 2. 3. 1. Gases consist of large numbers of tiny particles that are far apart relative to their size.2.~~

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