

## Calculations For A Level Chemistry Third Edition

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It consists of one carbon atom (Ar = 12) and two oxygen atoms (Ar = 16): Mr of CO<sub>2</sub> = 12 + 16 + 16 = 44. It could also be calculated this way: Mr of CO<sub>2</sub> = (1 × 12) + (2 × 16) = 12 + 32 = 44. Like ...

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Comprehensive mathematics foundation section. Work on formulae and equations, the mole, volumetric analysis and other key areas are included. Can be used as a course book as well as for exam practice.

Suitable for all examination specifications for students over 16, this friendly and reliable guide leads students through examples of each problem.

Many undergraduate students enter into chemistry courses from a wide range of backgrounds, often possessing various levels of experience with the mathematical concepts necessary for carrying out practical calculations in chemistry. *Chemical Calculations: Mathematics for Chemistry, Second Edition* provides a unified, student-friendly reference of mathematical concepts and techniques incorporated into the context of familiar chemical topics. Uniquely organized by chemical—rather than mathematical—topics, this book relates each mathematical technique to the chemical concepts where it applies. The new edition features additional, revised, and updated material in every chapter. It achieves greater clarity with newly improved organization of topics and cross-referencing where mathematical techniques occur more than once. The text also contains numerous worked examples along with end-of-chapter exercises and detailed solution—giving students the opportunity to apply previously introduced techniques to chemically related problems. An ideal course companion for chemistry courses throughout the length of a degree, the second edition of *Chemical Calculations: Mathematics for Chemistry* may also extend its utility as a concise and practical reference for professionals in a wide array of scientific disciplines involving chemistry.

Comprehensive mathematics foundation section. Work on formulae and equations, the mole, volumetric analysis and other key areas is included. Can be used as a course support book as well as for exam practice. Best-selling, experienced chemistry author.

Each topic is treated from the beginning, without assuming prior knowledge. Each chapter starts with an opening section covering an application. These help students to understand the relevance of the topic: they are motivational and they make the text more accessible to the majority of students. Concept Maps have been added, which together with Summaries throughout, aid understanding of main ideas and connections between topics. Margin points highlight key points, making the text more accessible for learning and revision. Checkpoints in each chapter test students' understanding and support their private study.

This Calculations Book \* provides plenty of practice in chemical arithmetic for students preparing for examinations at 16 plus \* covers all the main areas of the subject that involve calculations at this level \* gives concise summaries of all the relevant theory \* includes clear worked examples to illustrate each different type of calculation \* contains nearly 500 carefully graded questions, including many from recent examination papers \* includes a selection of multiple choice questions for further practice and revision

In this third edition of a textbook for A-Level chemistry, each topic starts at a level accessible to students who have attained Level 7/8 of the National Curriculum in Science, and is treated from the beginning without assuming that work from a previous course has been remembered.

*Chemistry in Quantitative Language*, second edition is an invaluable guide to solving chemical equations and calculations. It provides readers with intuitive and systematic strategies to carry out the many kinds of calculations they will meet in general chemistry.

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