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the "Big Ideas" in AP Chemistry : The Following PowerPoints were not created by me, but rather, were the collaborative effort of many AP Chemistry teachers. You may well find them useful. They only "work" correctly if you view them on a computer or tablet running PowerPoint. Chapter 2,3,7,8 PowerPoint Review (old Big Idea #1) handout

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AP WORKSHEET 15a: ANSWERS 1. (a) Rate = k [H<sub>2</sub>]<sup>2</sup>[Br<sub>2</sub>]<sup>0</sup> 1.92 x 10<sup>-3</sup> M<sup>-1</sup> s<sup>-1</sup> (b) 2nd order, (2 + 0 = 2) 2. (a) Rate = k [O<sub>2</sub>]<sup>2</sup> [NO]<sup>2</sup> (b) 500 M<sup>-2</sup> s<sup>-1</sup> 3. (a) (mol L<sup>-1</sup>)<sup>-2</sup> s<sup>-1</sup> (b) M<sup>-1</sup> min<sup>-1</sup> (c) g s<sup>-1</sup> There are, of course, other ways to express these units correctly 4. Start with known concentrations of the two reactants and carry out the reaction.

Revised August 2009 - Weebly

Mixing 1000 mL of an aqueous solution with 1000 mL of water represents a A; Barton Community College; CHEM n/a - Spring 2011

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Solved: Adrian Dingle's Chemistry Pages And ChemEducator L ...

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