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Chemistry: 7.1-7.6

Equilibrium,

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9.1-9.3, 9.5, 7.14

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□ Write the

equilibrium

expression. Look

up the value for K_a

(in a table). □ Write

down the initial

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and equilibrium
concentrations for
everything except
pure water. □ We
usually assume
that the
equilibrium
concentration of
 H^+ is x .

□ Substitute into the
equilibrium
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expression and
solve. □ Remember

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to convert x to pH
if necessary.

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Ksp. Students
should be able to...

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in equilibrium .

Find the reaction .
quotient, Q , (K_{eq})

for a reaction
equation . Solve for

Q given . initial
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Chemistry is a
subject that builds
on itself from the
ground up. If you
don't understand...

#2: Do It Yourself.

When studying a
subject that
requires step-by-
step problem

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Solving, students

often read

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answer... #3:

Double Check for
Logic and ...

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topics include

atomic theory,

chemical bonding,

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phases of matter,
solutions, types of
reactions,

equilibrium,

reaction kinetics,

and

thermodynamics.

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take an AP

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should take at least
two years of high
school math before
enrolling in this

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UNIT 1: Atomic
Structure and
Properties

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Description,
Effective Fall 2020
The Chemistry of
Acids & Bases 4

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STRENGTHS OF

ACIDS AND BASES

Strength is determined by the position of the "dissociation" equilibrium. $\frac{3}{4}$

Strong acids/strong bases 1.

dissociates completely in water 2. have very large dissociation

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Equilibrium
or K values $\frac{3}{4}$

Weak acids/weak
bases 1. dissociate
only to a slight
extent in water 2.

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Chemistry of Acids
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Chemistry. Unit:

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equilibrium.

Lessons.

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Reactions in

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equilibrium (Opens
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Calculating
equilibrium

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Small x

approximation for
small ...

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Notes. These notes
come from several

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are in-depth,
others give a broad
overview. Some
focus more on
notes, others on
working through
practice questions.
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you a
comprehensive
guide of what you
need to know for
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the College Board
Website The AP
Chemistry course is
designed to be the
equivalent of the
general chemistry
course usually
taken during the
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